

# Emission-reduction catalysts to enable use of highly efficient CIDI engines



## O A A T A C C O M P L I S H M E N T S

### Advanced Catalysts for Clean CIDI Engines

#### Challenge

Because compression-ignition direct-injection (CIDI) engines have the highest thermal efficiencies of any proven automotive power plant, they are seen as being among the most promising candidates to deliver fuel economies of up to 80 miles per gallon in vehicles being developed under the Partnership for a New Generation of Vehicles Program. Before widespread use of CIDI engines can become a reality, however, their nitrogen oxides ( $\text{NO}_x$ ) emissions must be reduced to meet federal Tier 2 standards, which will go into effect in 2004.



*Full-development size (110 cubic inch) catalytic converter in take-apart can assembly.*

#### Technology Description

To overcome this emissions barrier, the U.S. Department of Energy (DOE), DaimlerChrysler Corporation, Ford Motor Company, and General Motors Corporation are working together to develop advanced catalytic materials for catalytic converter systems that reduce  $\text{NO}_x$  emissions from CIDI engines. Research has taken two complementary directions. One involves development of new zeolite-based catalysts that promise to be effective in  $\text{NO}_x$  removal at high temperatures, while the other focuses on development of platinum (Pt)-based hydrous metal oxide (HMO)-supported catalysts that operate at low temperatures.

procedures were optimized to achieve at least 80%  $\text{NO}_x$  conversion at high temperatures (250-450° C). However, catalyst exposure to water in the exhaust stream was found to degrade  $\text{NO}_x$  conversion at temperatures below 400° C.

Researchers at Sandia National Laboratories (SNL) achieved 60%  $\text{NO}_x$  conversion with the Pt-based/HMO catalysts. This conversion efficiency was not sufficient to meet Tier 2 standards. Subsequent SNL research led to an understanding of the limitations of supported Pt catalysts for  $\text{NO}_x$  removal in CIDI engines.

#### Benefits

Substantial  $\text{NO}_x$  conversion efficiencies have been achieved with advanced catalytic materials.

At the same time, the need for a better approach to  $\text{NO}_x$  control has been highlighted.

#### Accomplishments

Researchers at Los Alamos National Laboratory (LANL) discovered a new family of Ferrierite (FER) zeolite-based catalysts. Zeolite support materials and catalyst pretreatment

#### Contacts

Kathi Epping  
Manager, Diesel  
Combustion and Emission  
Control R&D  
202-586-7425  
202-586-9811 fax  
kathi.epping@hq.doe.gov

Timothy J. Gardner  
Sandia National  
Laboratories  
505-272-7621  
505-272-7336 fax  
tjgardn@sandia.gov

## Commercialization

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A number of patents resulted from the research. Patents were issued to SNL for NO<sub>x</sub>-adsorbing materials and NO<sub>x</sub>-reduction catalysts. LANL filed for a patent on the FER-supported metal catalysts.

SNL also developed a process to transfer technology to its industrial research partners and their designated catalyst suppliers.

## Awards

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1997 Vice President's PNGV Award

1999 National Laboratory CIDI R&D Award (SNL)

2000 National Laboratory CIDI R&D Award (LANL)

## Future Activities

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To achieve a NO<sub>x</sub> conversion efficiency that meets the Tier 2 standard, SNL's NO<sub>x</sub>-reduction reactor will be refitted to support a more promising approach, selective catalytic reduction of NO<sub>x</sub> by urea and ammonia. Mixed catalyst applications will also be pursued to increase the NO<sub>x</sub> conversion efficiency of lean-burn catalysts.

The effect of water on FER-supported metal catalysts will be explored further. The possible effects of particulates on the NO<sub>x</sub>-reduction performance of lean-burn NO<sub>x</sub> catalysts will also be investigated.

## Partners in Success

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- DaimlerChrysler Corporation
- Ford Motor Company
- General Motors Corporation
- Los Alamos National Laboratory
- Oak Ridge National Laboratory
- Sandia National Laboratories

